

The bright side of defects: Chemistry and physics of persistent and storage phosphors

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law
e



1945 vs. nowadays





1702

the Beginning

of the University

of Wrocław



The Bull establishing the University

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University just after the World War II



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University Nowadays



University Nowadays



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Graduation



Faculty of Chemistry

Researchers	~110
PhD Students	~100
Students	~600



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Luminescent Materials Research Group



The nice part
is over



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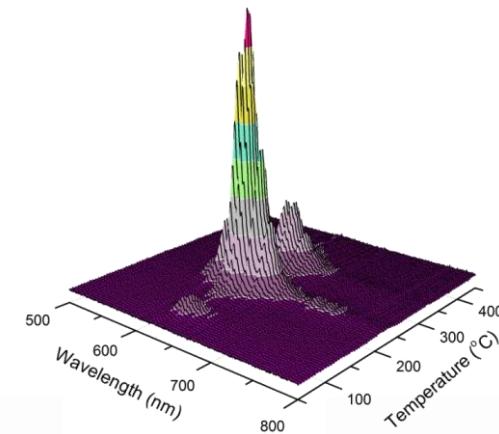


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Colin Humphreys

“Crystals are like people,
it is the defects in them
which tend to make them interesting!”



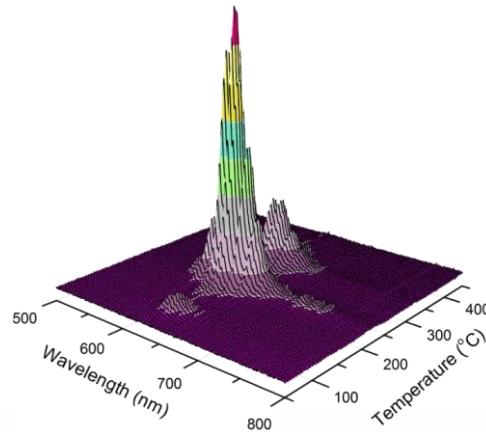
Defect Types a simplified approach

Intrinsic defects: Schottky and Frenkel defect (thermally created)

Extrinsic defects: Impurities (intentional or unintentional dopants)

And these two lines can be easily developed into:

- An article
- A book chapter
- A book



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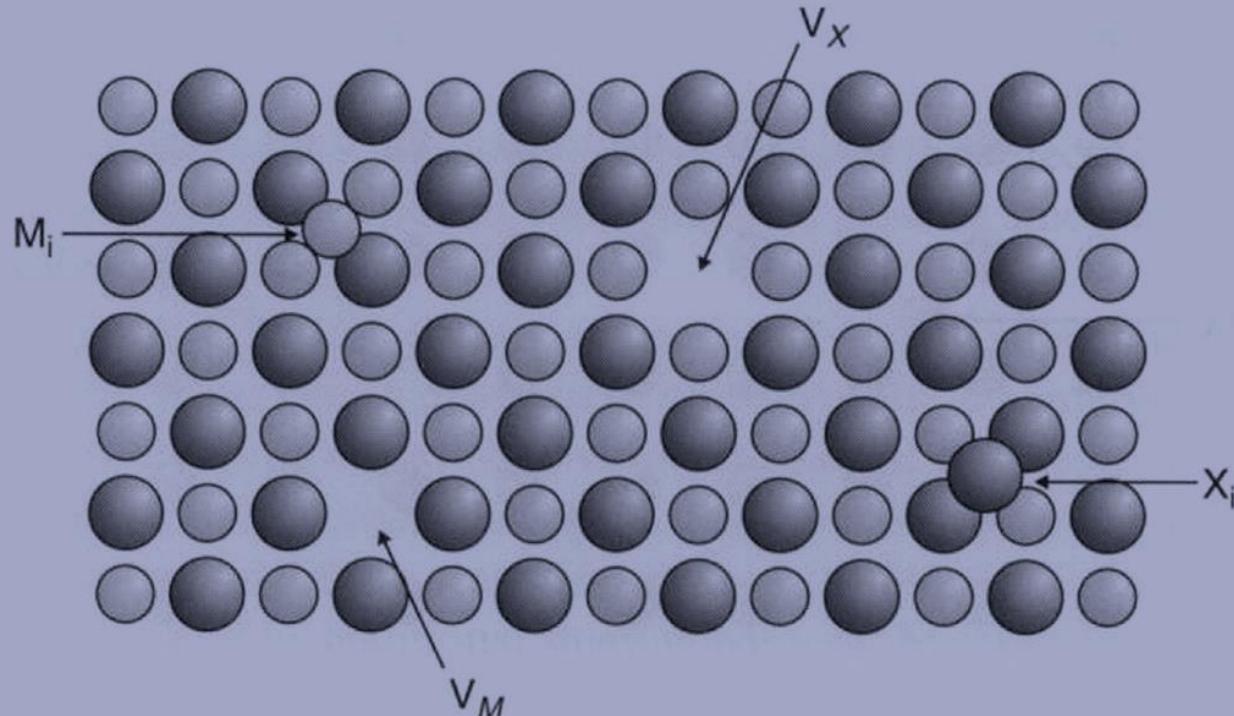


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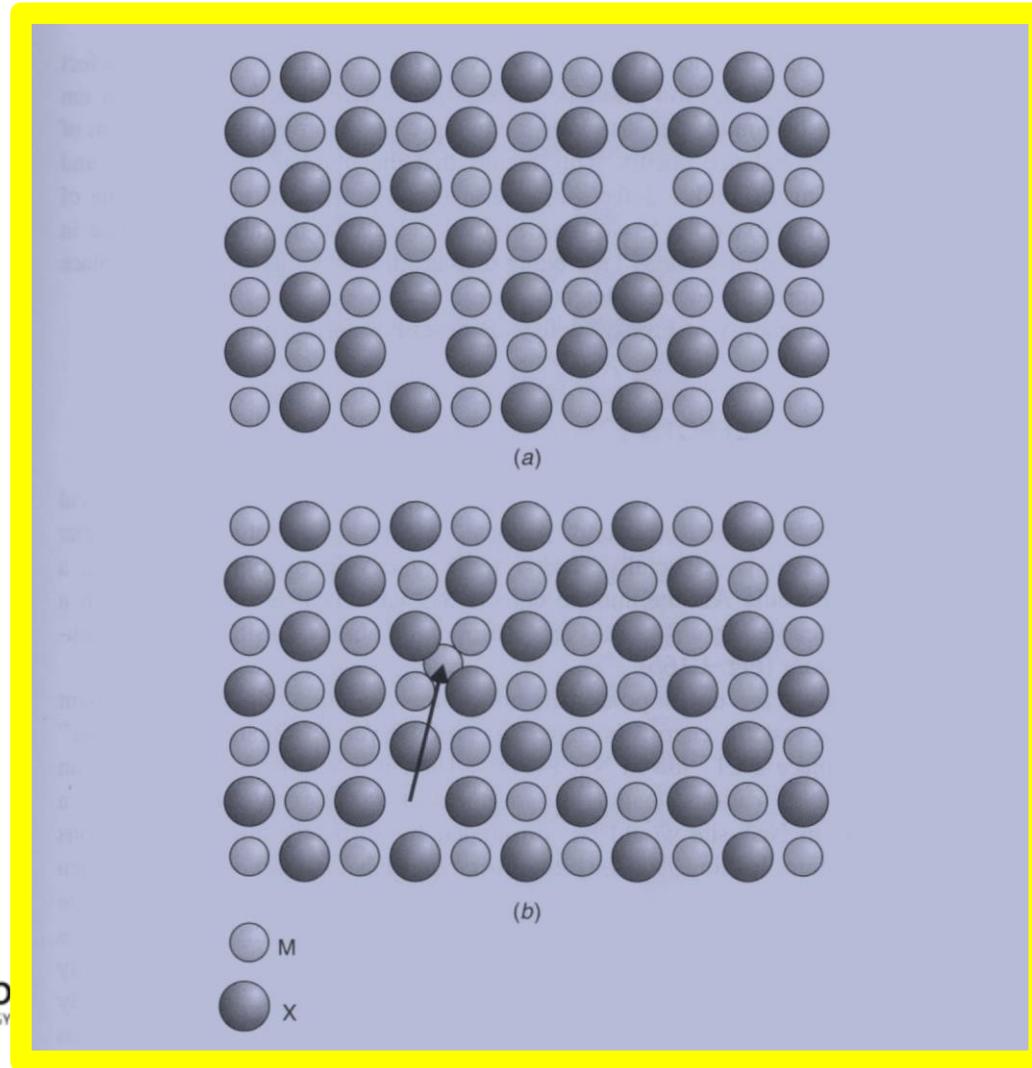
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Types of Defects in Crystalline Solids



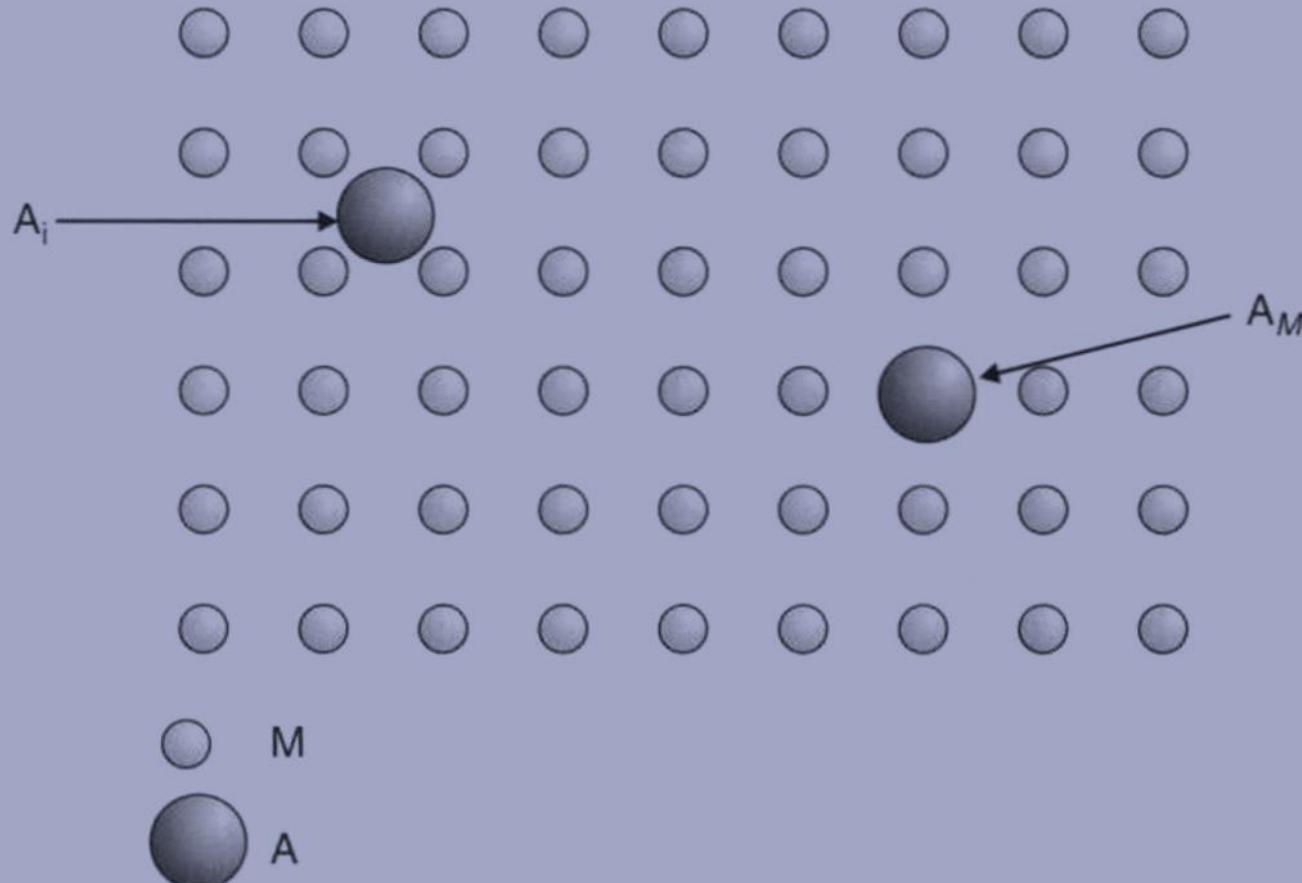
Pair Defects: Schottky and Frenkel Defects. Their Populations Increases with Temperature



Schottky

Frenkel

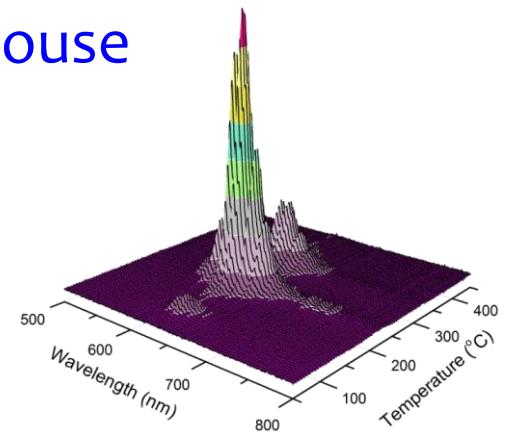
Impurity Point Defects



Erice, Sicily, Italy, 2014

**Andries Meijerink: „We, who research luminescence,
need to learn more about the NATURE of DEFECTS”**

During the discussion at
Prof. Baldassare (Rino) Di Bartolo's Family House



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What happens in Storage (and persistent) phosphors?

Defects
play their game



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A story on

$\text{Lu}_2\text{O}_3:\text{Tb,M};$
 $\text{Lu}_2\text{O}_3:\text{Pr,M};$

M=Sr/Ca, Hf, Ti, Ta, Nb



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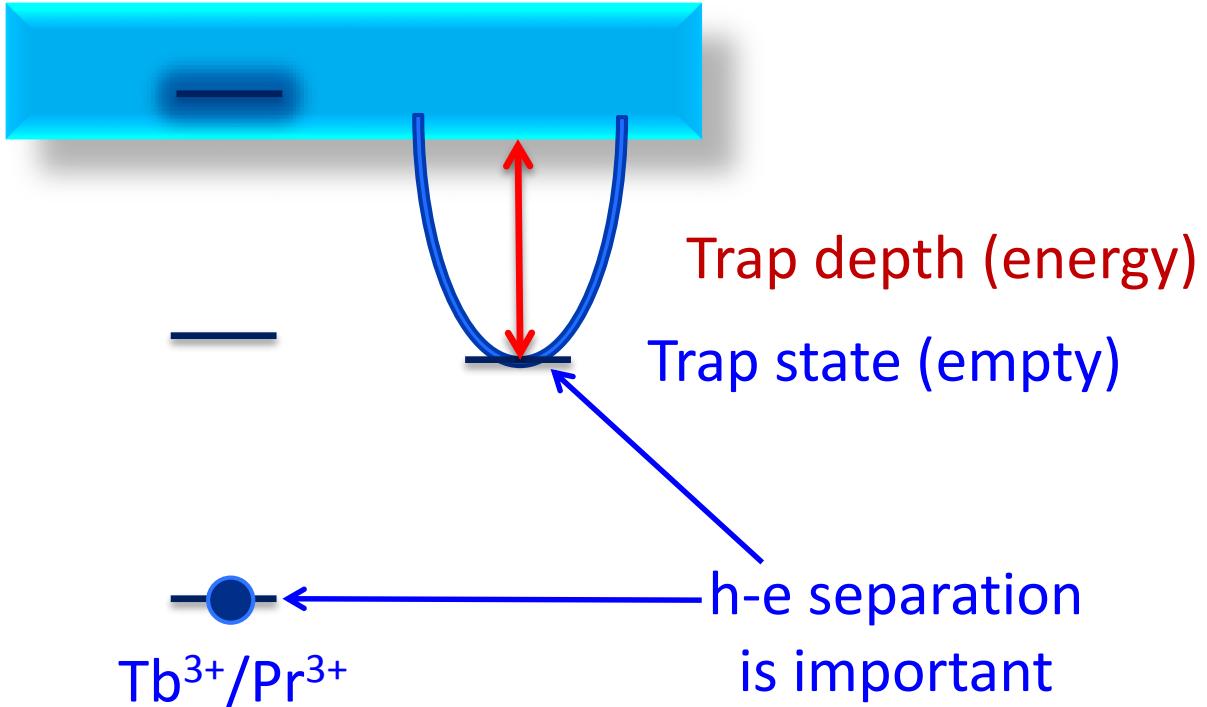


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Energy storing = immobilizing carriers in excited states



Energy releasing

Thermoluminescence



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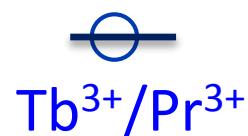
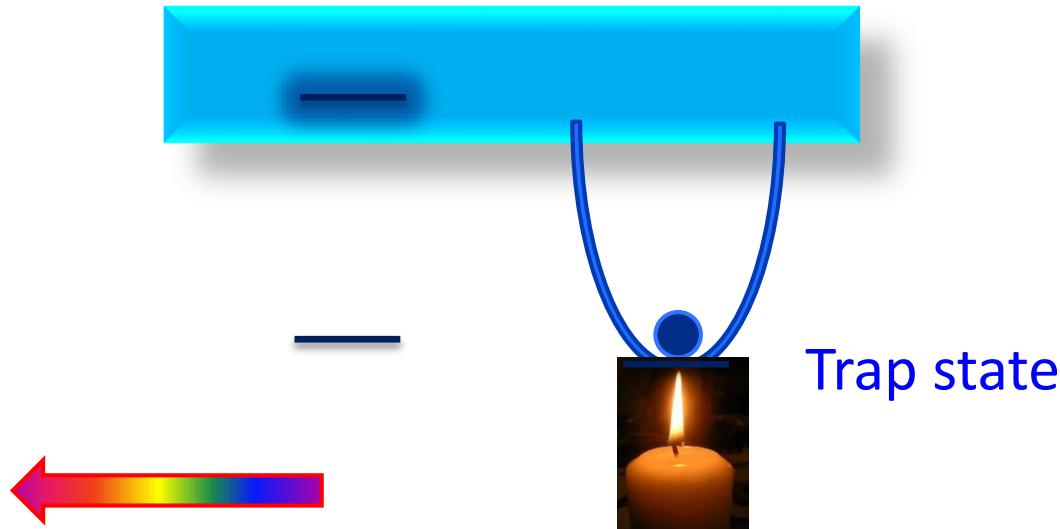


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Energy releasing - thermoluminescence



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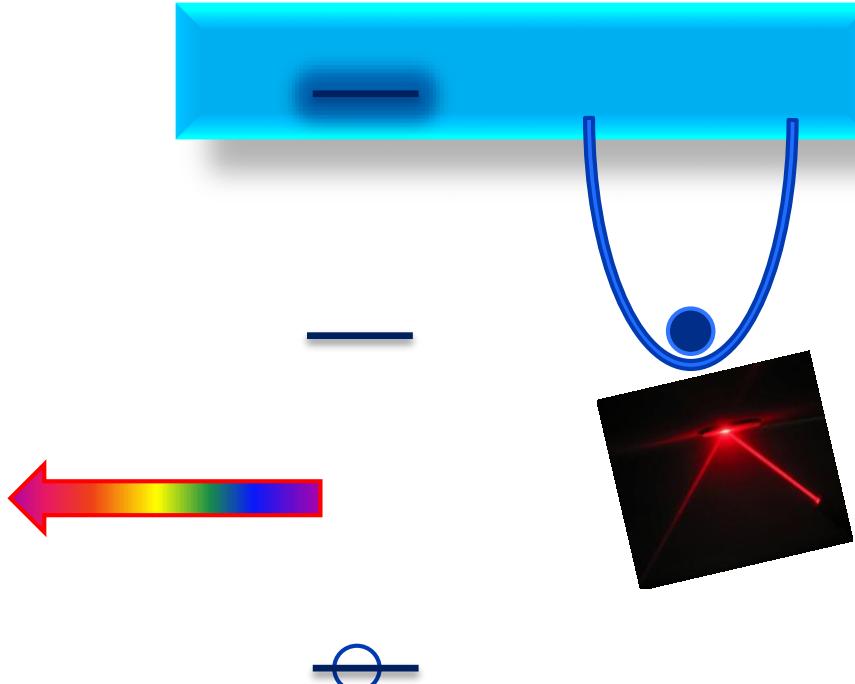


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Energy releasing Optically Stimulated Luminescence



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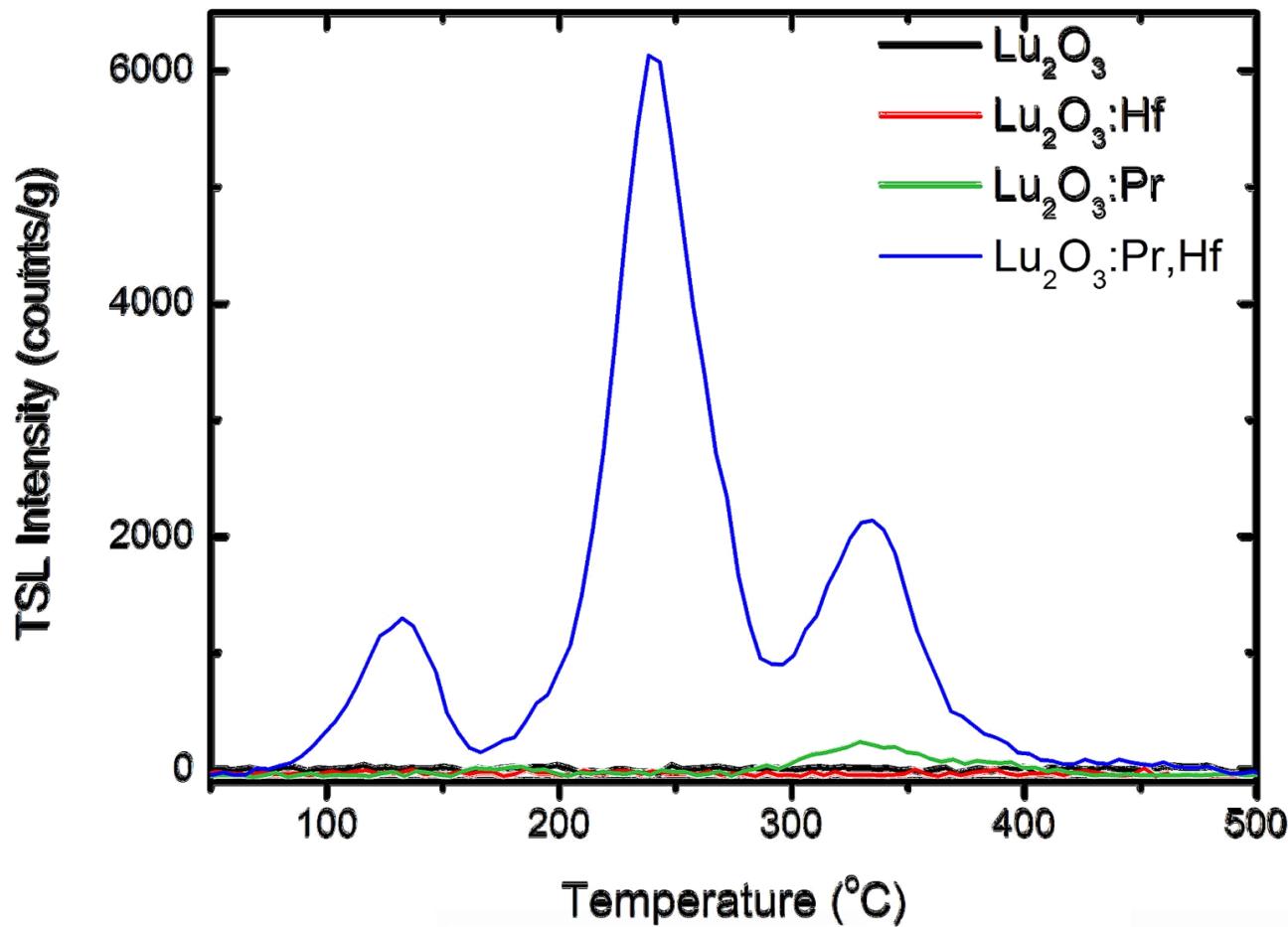
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TL – a combined effect of dopants

$\text{Lu}_2\text{O}_3:\text{Pr},\text{Hf}$



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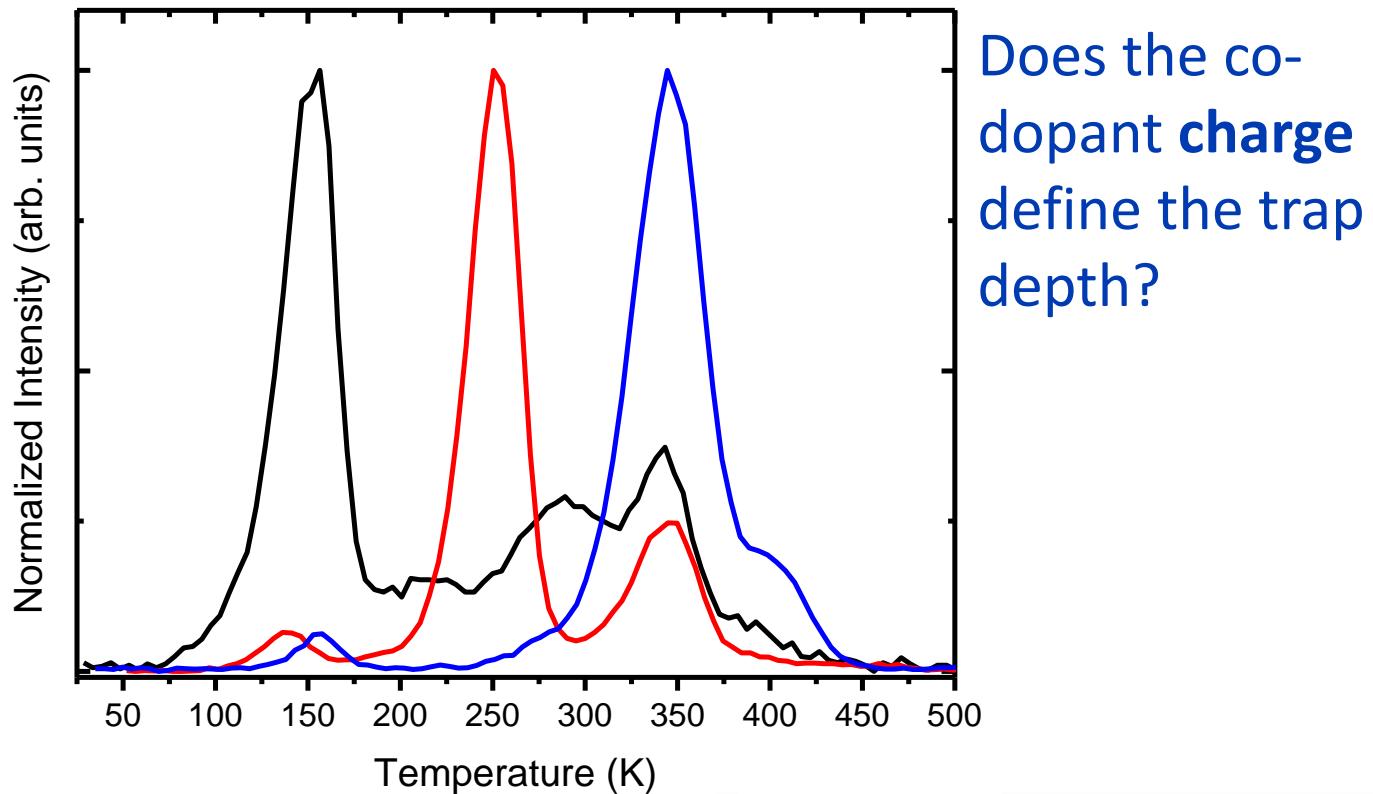


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$\text{Lu}_2\text{O}_3:\text{Pr},\text{M}$ - co-dopant define the trap depth(s)?

$\text{Sr}(\text{II}) < \text{Hf}(\text{IV}) < \text{Nb}(\text{V})$



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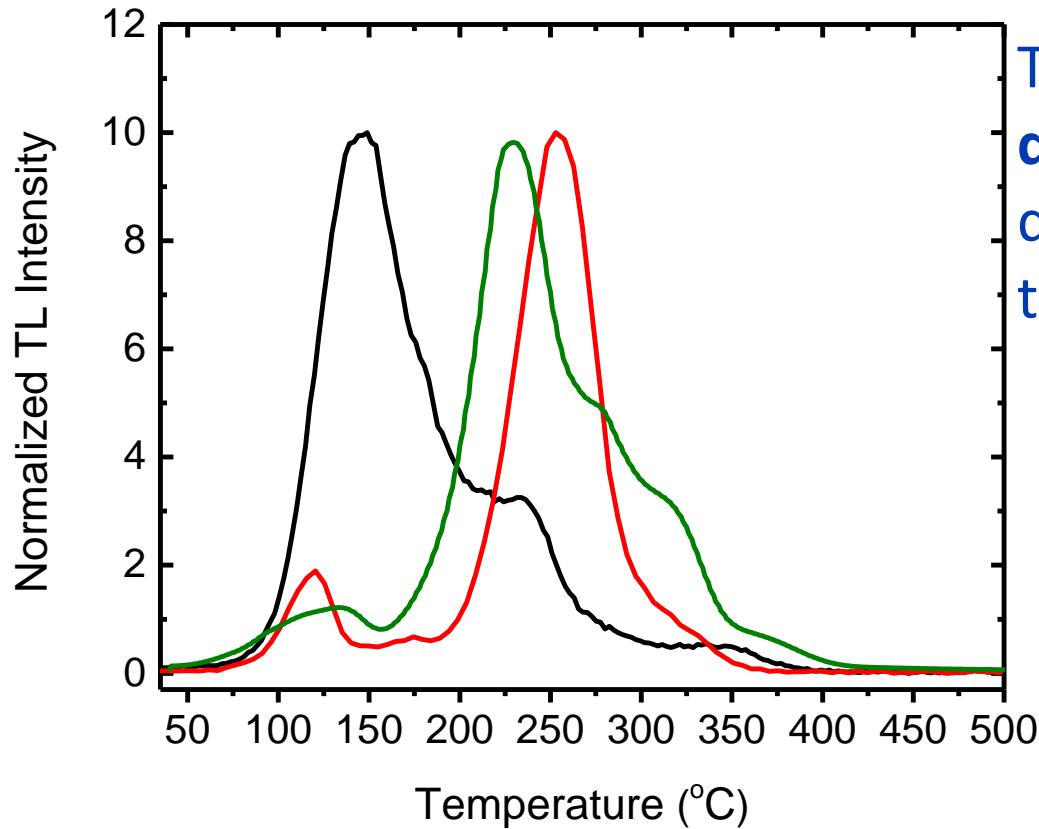
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Lu₂O₃:Tb,M - co-dopant and the trap depth(s)

!!!

Ca(II) < Nb(V) < Hf(IV)



The charge
does not
directly define
the trap depth



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Comparison of TL **Lu₂O₃:Pr,M vs. Lu₂O₃:Tb,M**



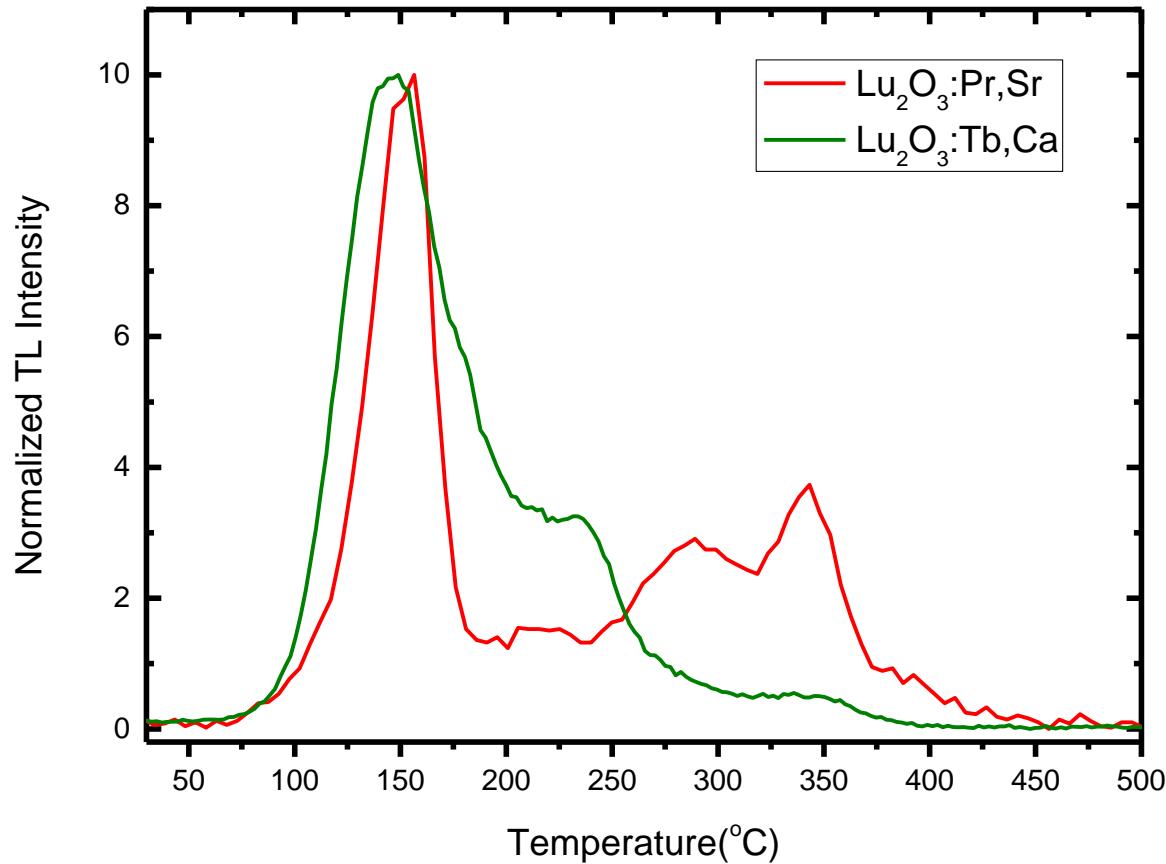
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Pr,Sr vs. Tb,Ca



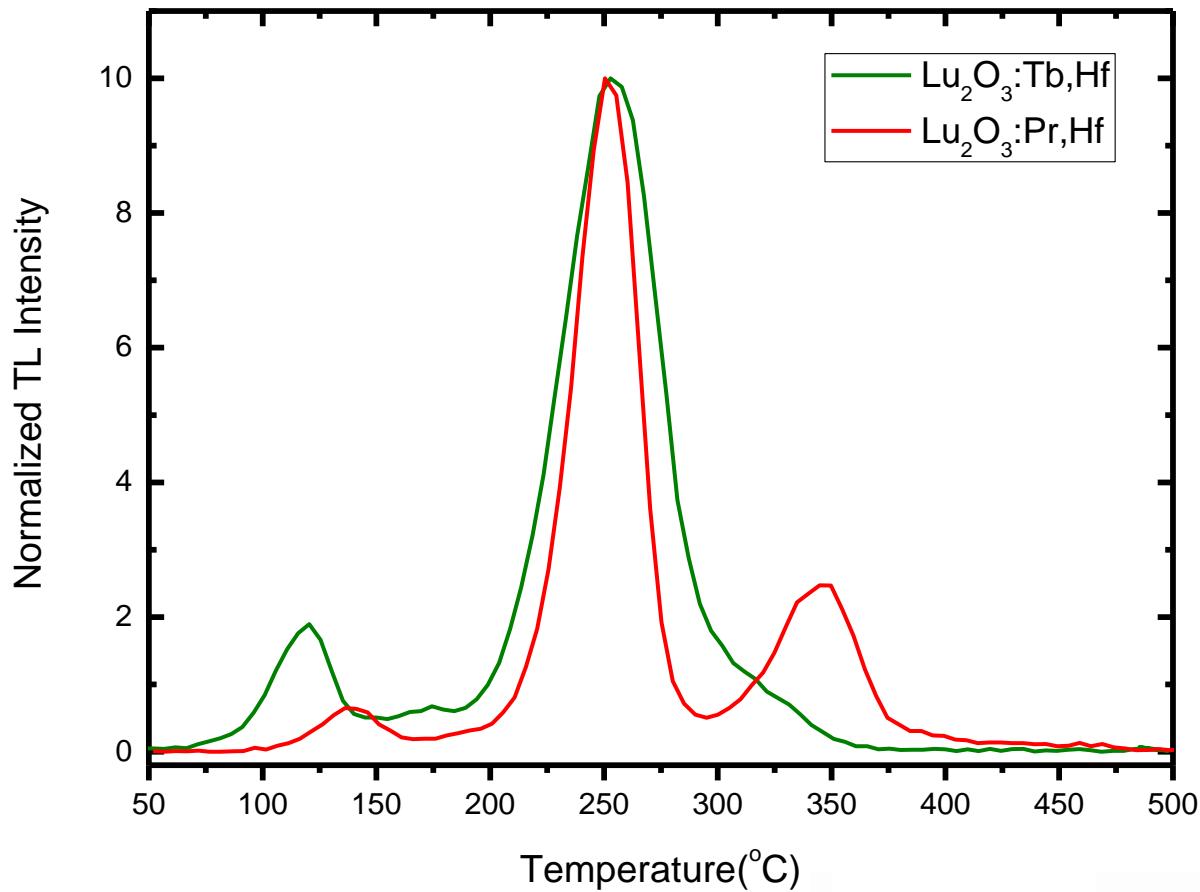
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Pr,Hf vs. Tb,Hf



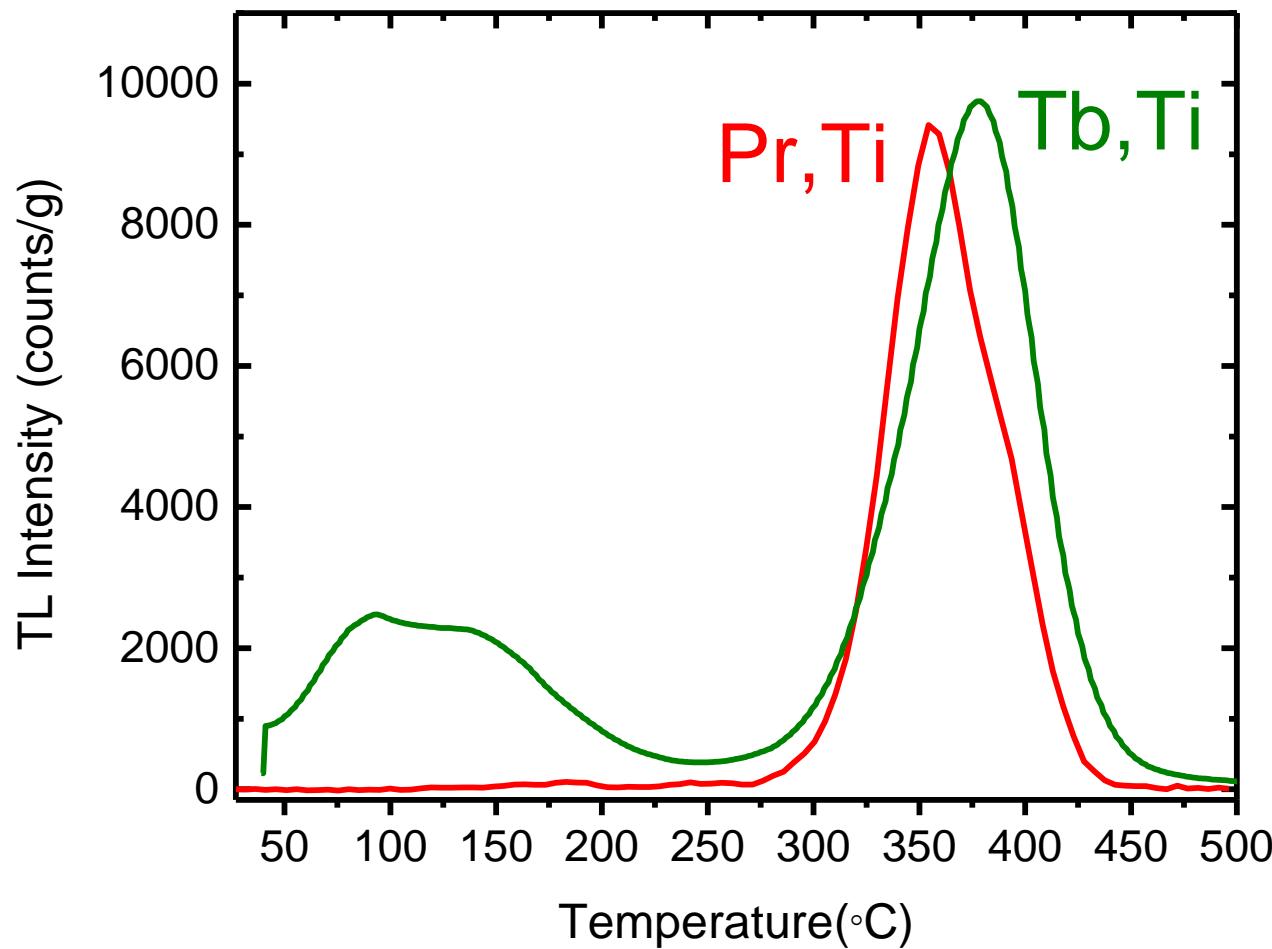
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Pr,Ti vs. Tb,Ti



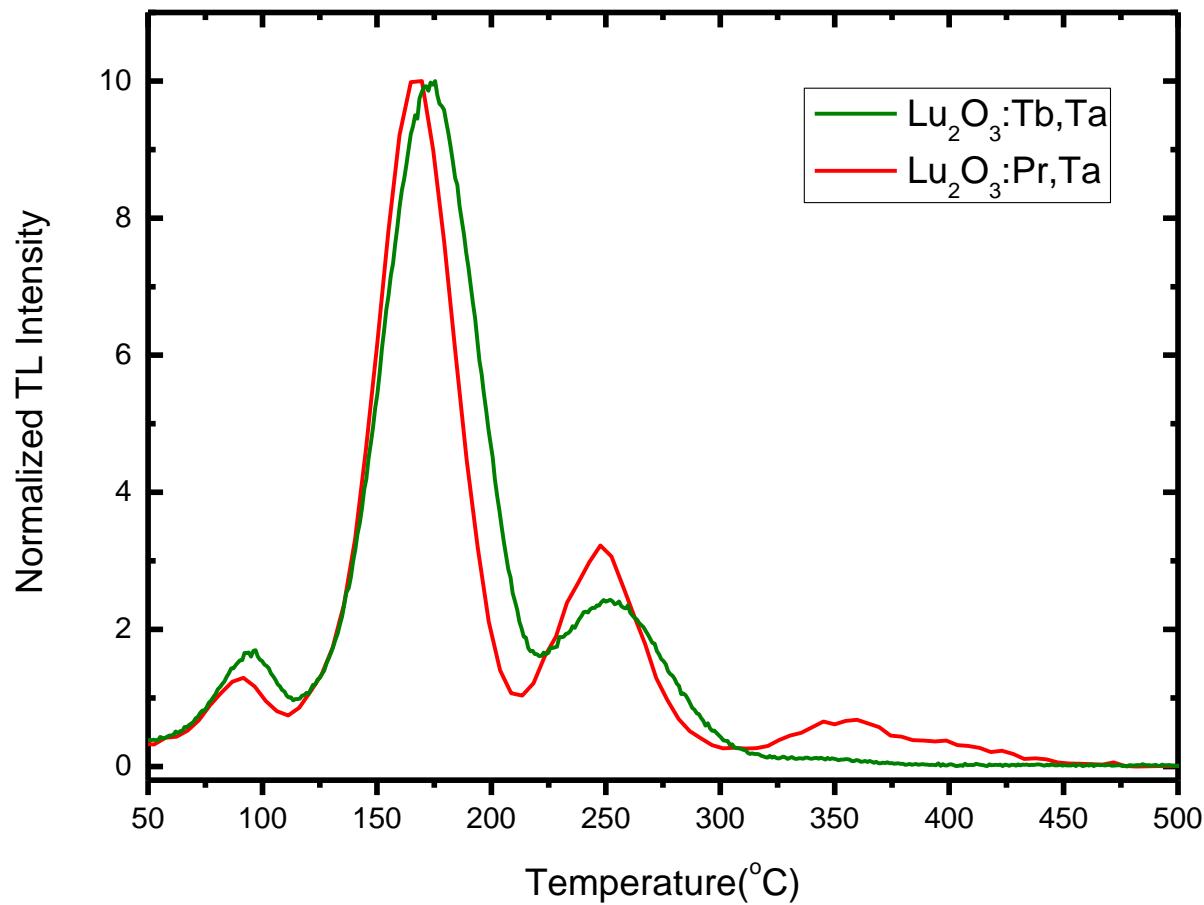
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Pr,Ta vs. Tb,Ta



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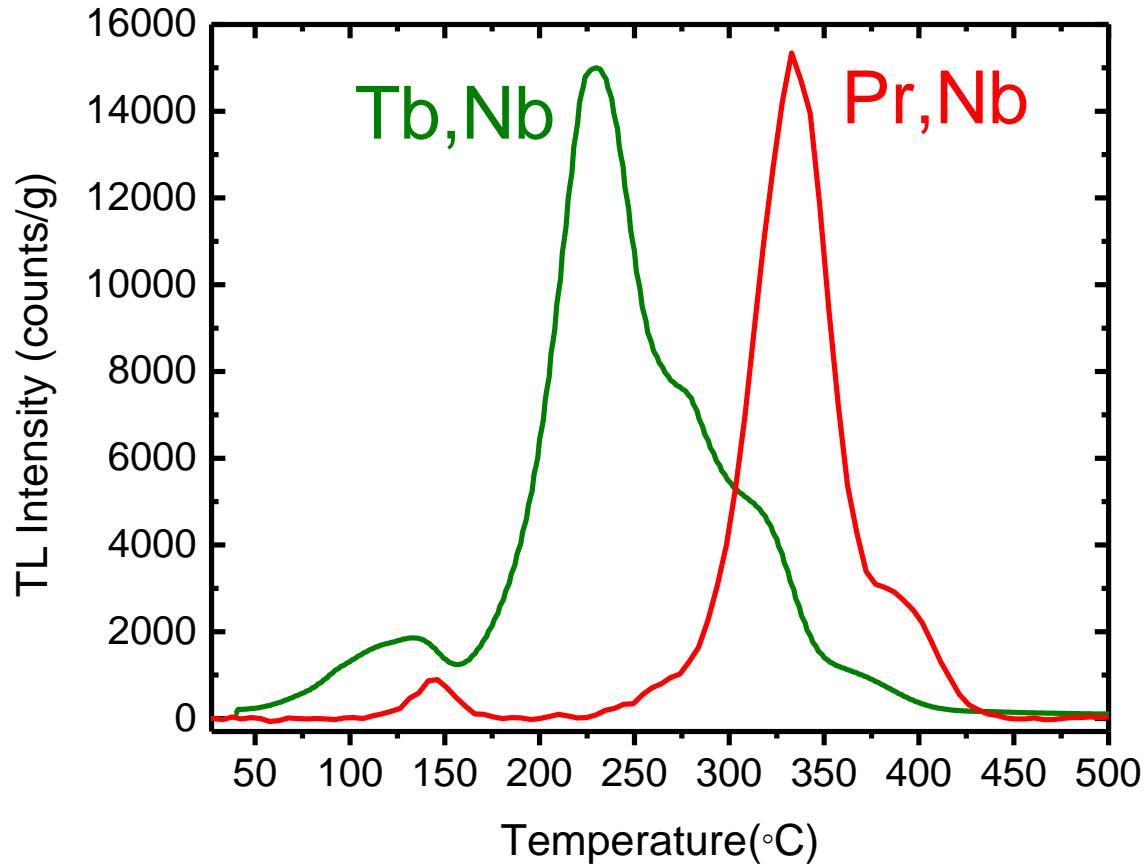


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Pr,Nb vs. Tb,Nb

?



?



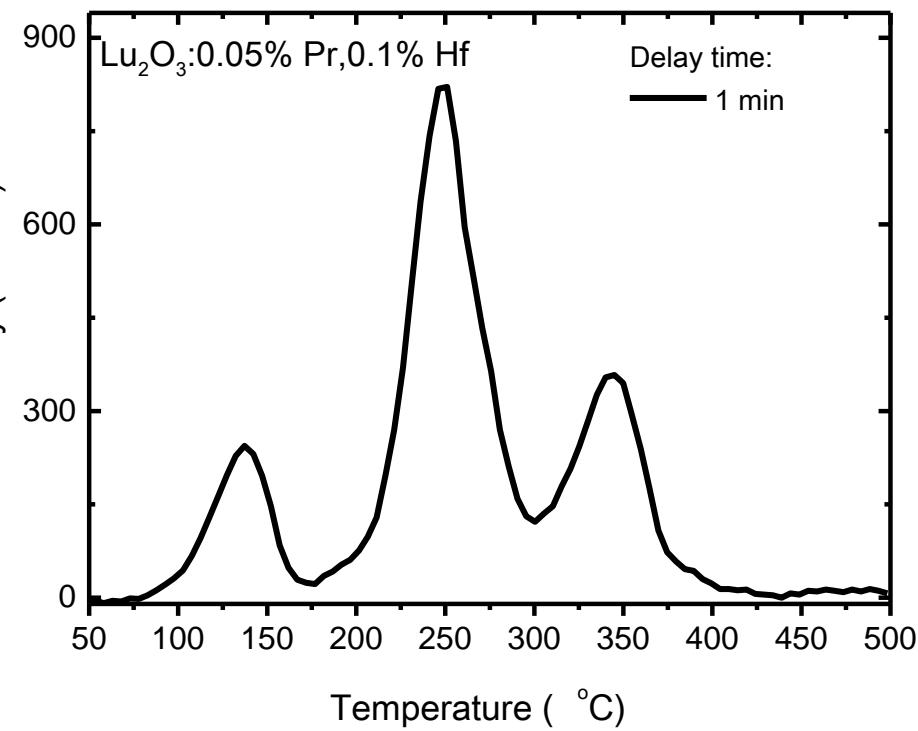
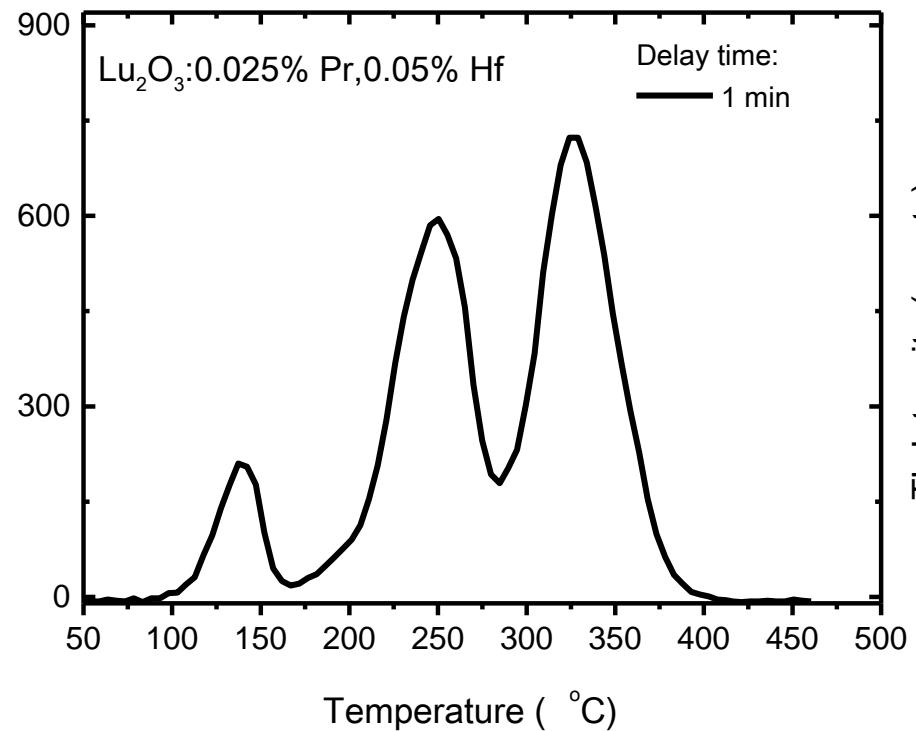
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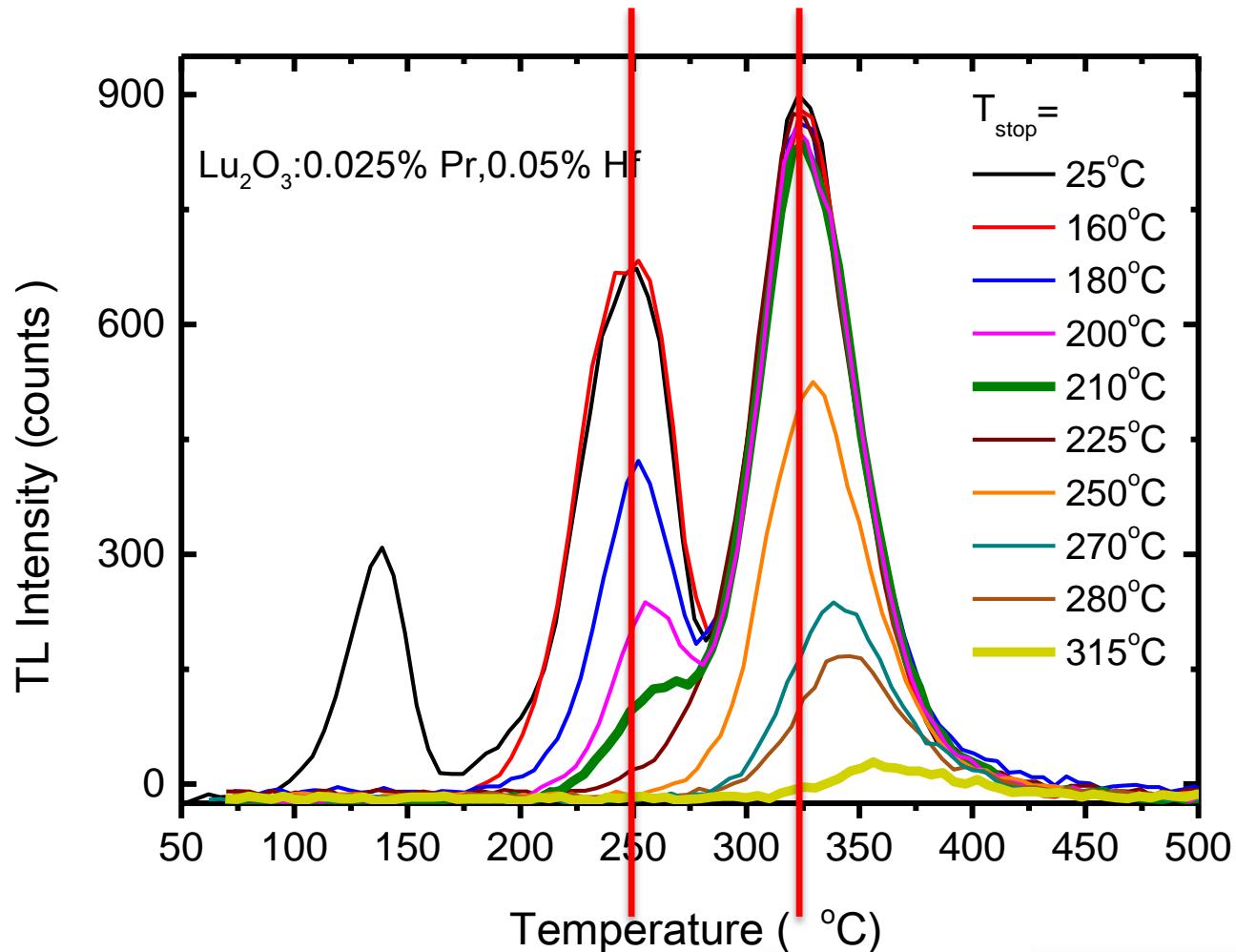
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Concentrations are very important



Thermal „partial cleaning” reveals the traps’ structure



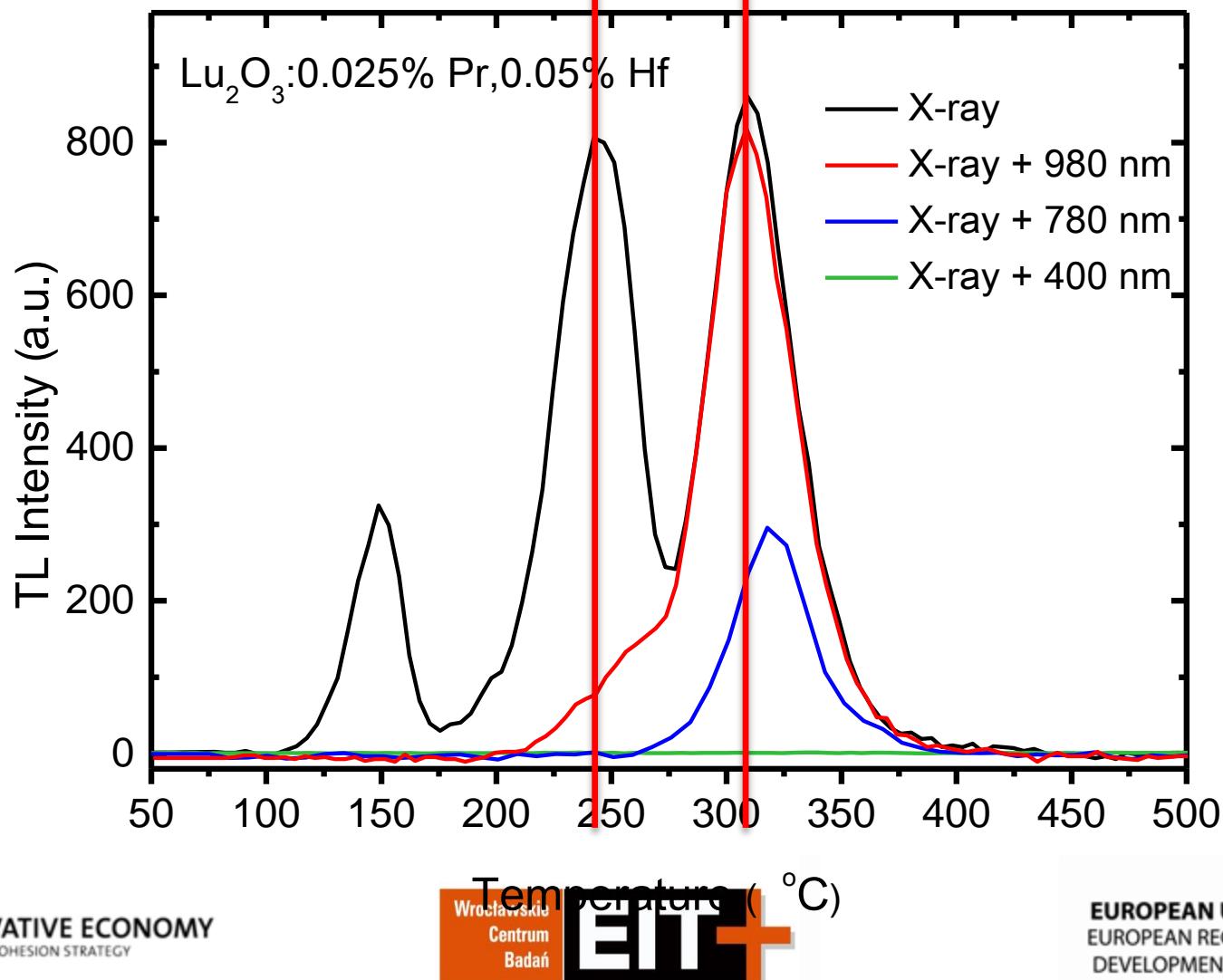
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Optical „partial cleaning” reveals the traps structure



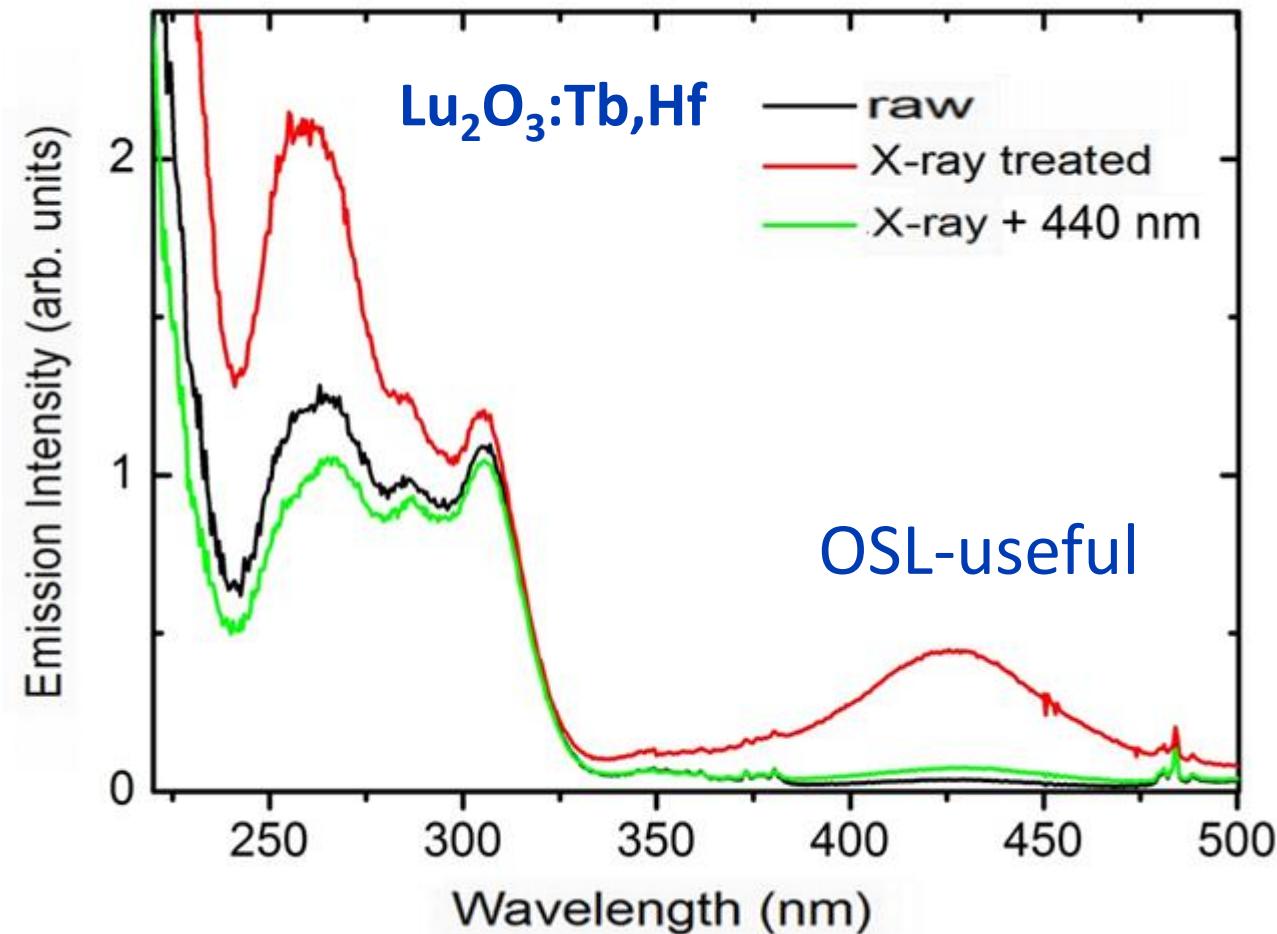
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Changes are seen in excitation spectra and are reversible



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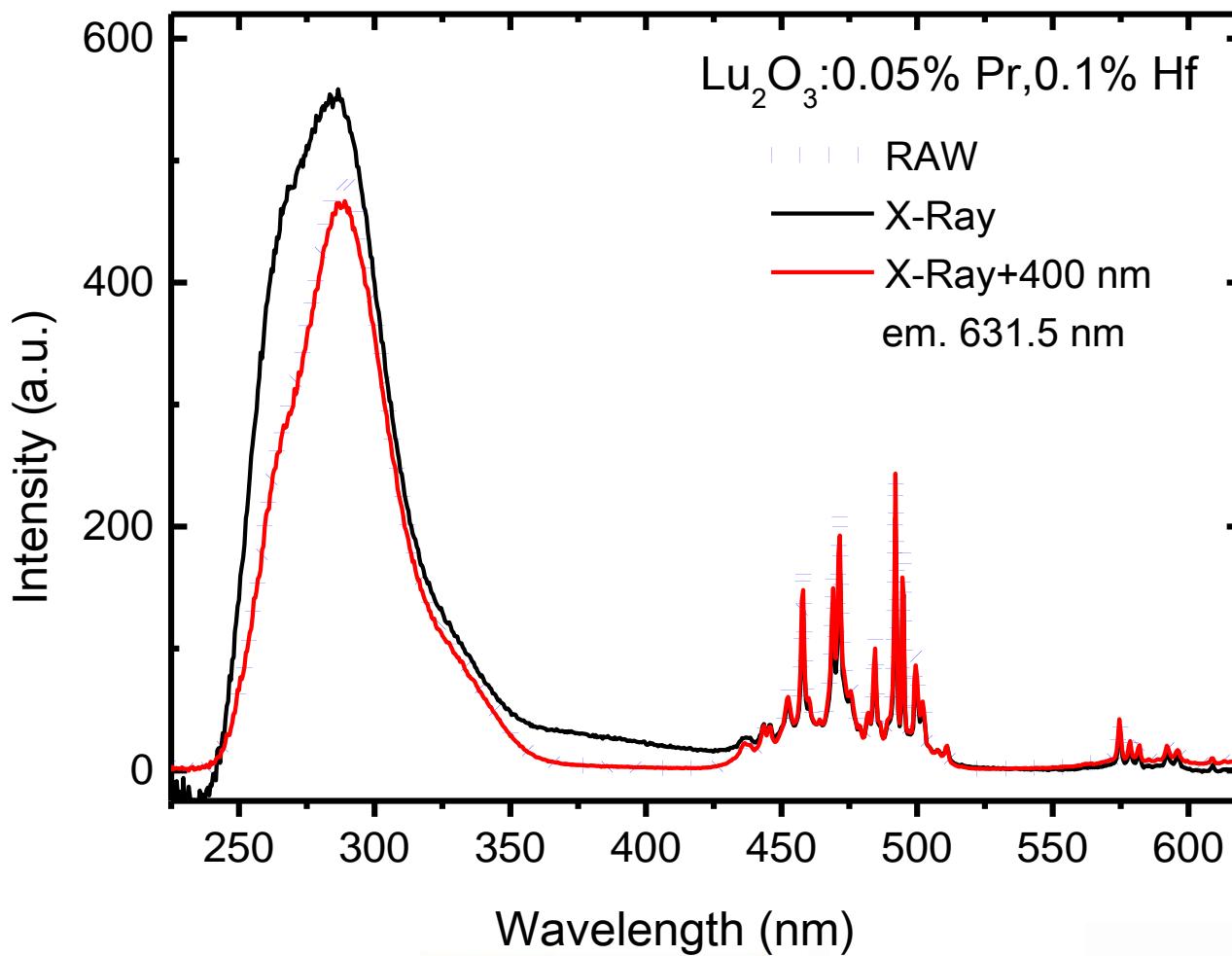


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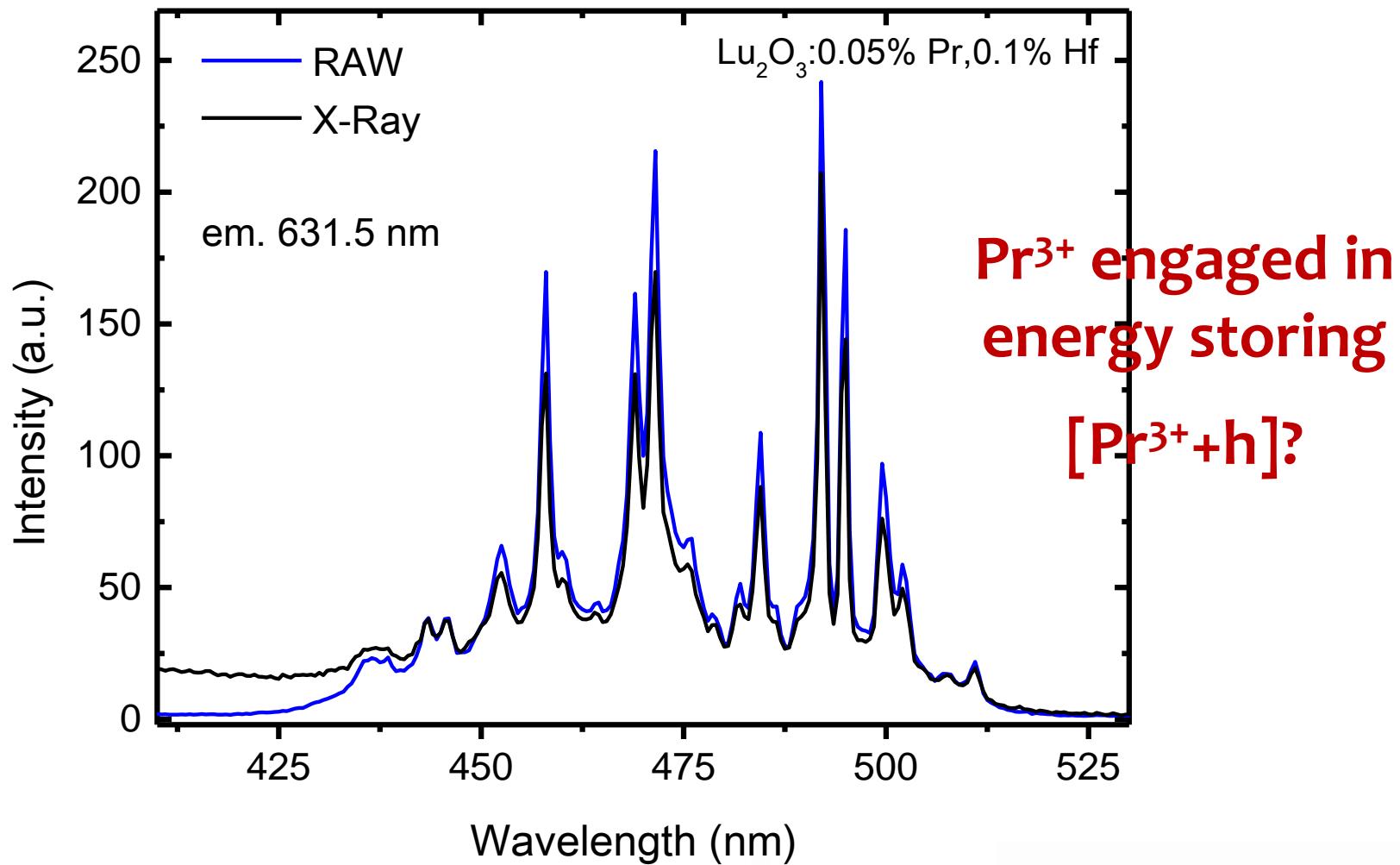
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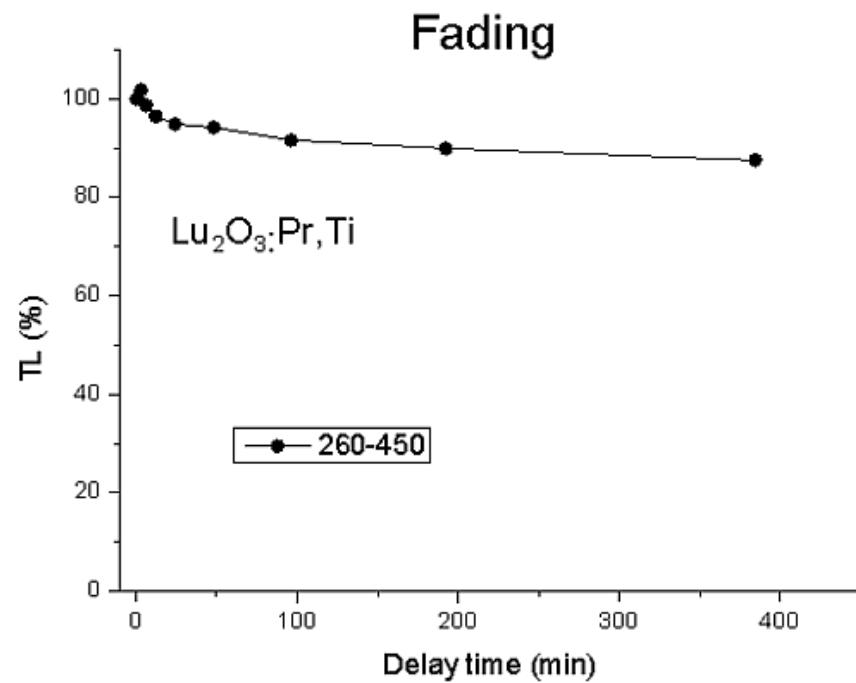
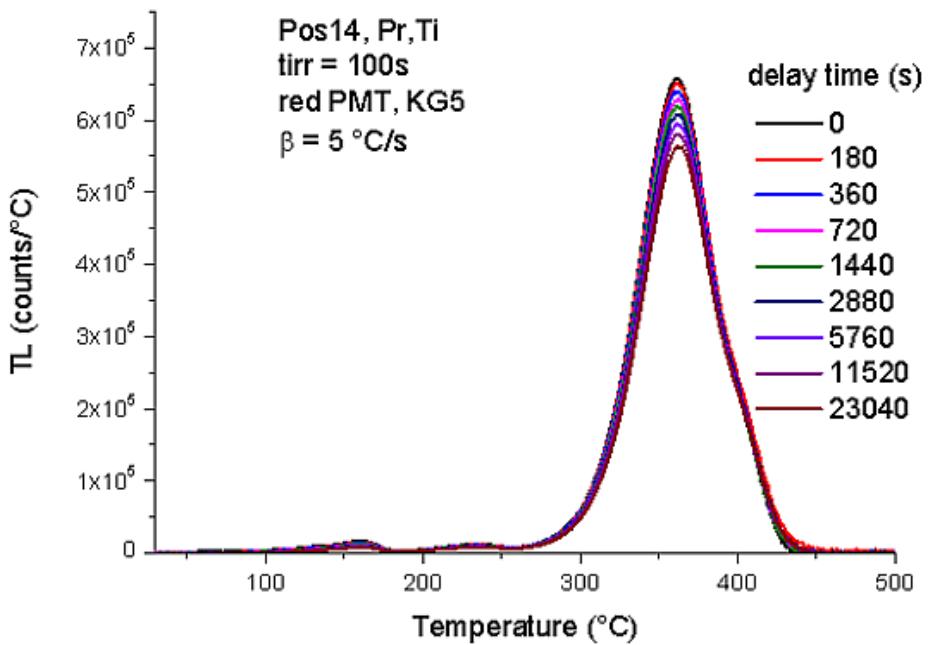
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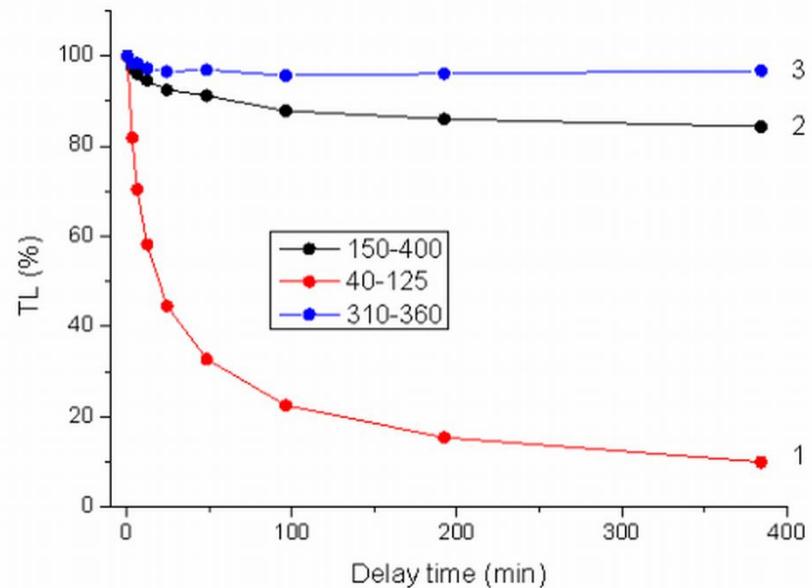
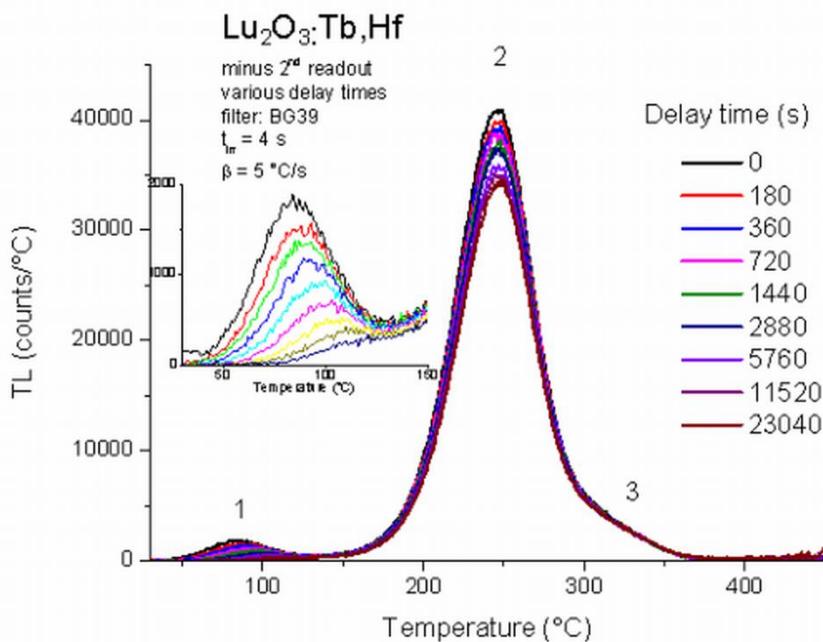
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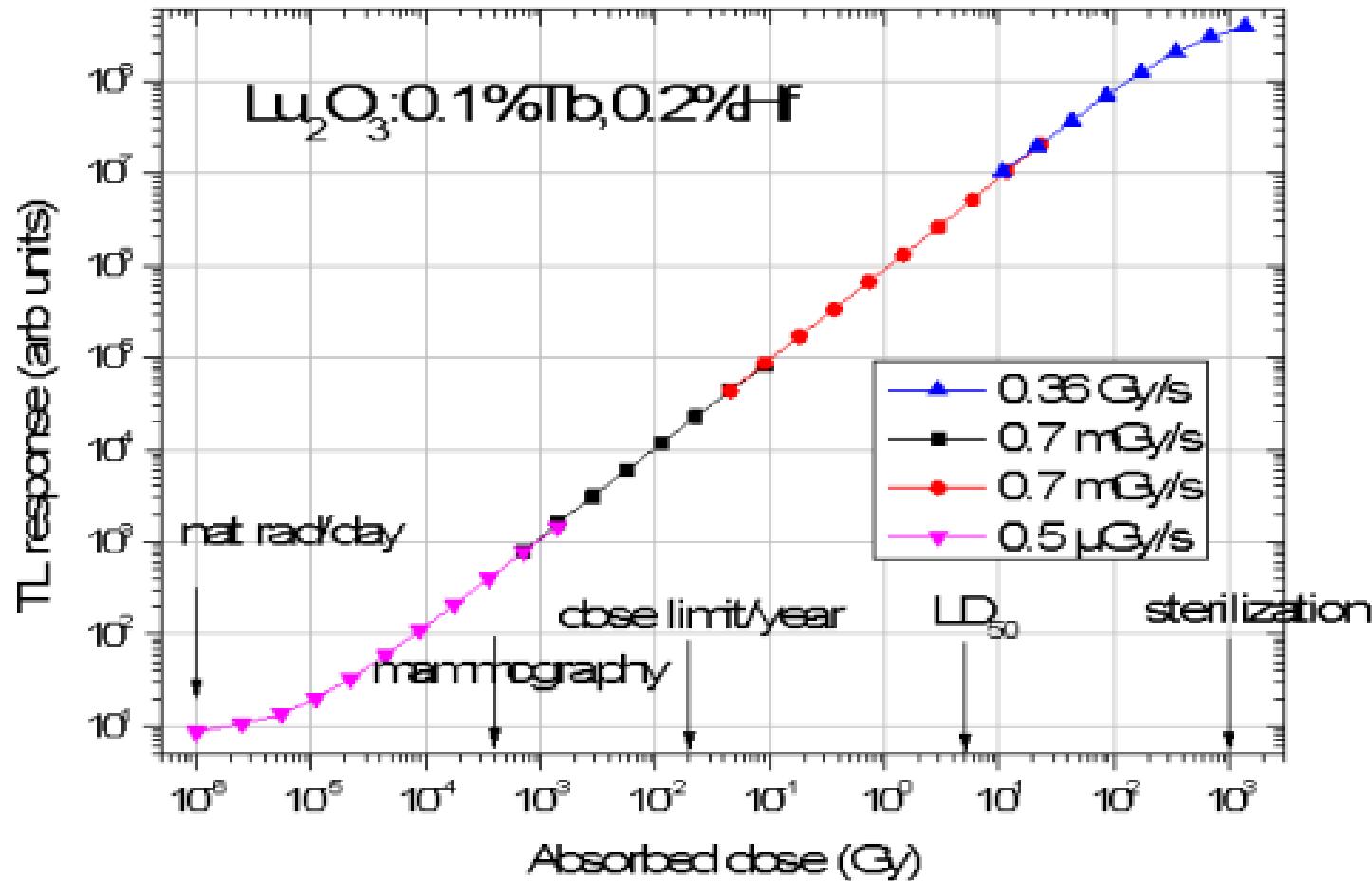
Lu₂O₃:Pr,Ti – fading



$\text{Lu}_2\text{O}_3:\text{Tb,Hf}$ – fading



Lu₂O₃:Tb,Hf – range of „linear” response 7 orders of magnitude – exceptional



environmental, personal, radiotherapy, sterilization

Summary

- Two families of storage phosphors were discovered:
green-emitting Lu₂O₃:Tb,M and red-emitting Lu₂O₃:Pr,M
- Response - dose „linear” dependence covers >7 orders of magnitude
- No radiation damage observed up to at least 1 kGy range of doses



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and Dr. Adrie Bos (Delft TU)



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Thank you

